

Fluids and Solids & Liquid and Gases

In the above we have discussed the differences between the behaviour of solids and fluids under an applied force. Summarising, we have;

- ✚ For a **solid** the strain is a function of the applied stress (providing that the elastic limit has not been reached). For a **fluid**, the rate of strain is proportional to the applied stress.
- ✚ The strain in a **solid** is independent of the time over which the force is applied and (if the elastic limit is not reached) the deformation disappears when the force is removed. A **fluid** continues to flow for as long as the force is applied and will not recover its original form when the force is removed.

It is usually quite simple to classify substances as either solid or liquid. Some substances, however, (e.g. pitch or glass) appear solid under their own weight. Pitch will, although appearing solid at room temperature, deform and spread out over days - rather than the fraction of a second it would take water.

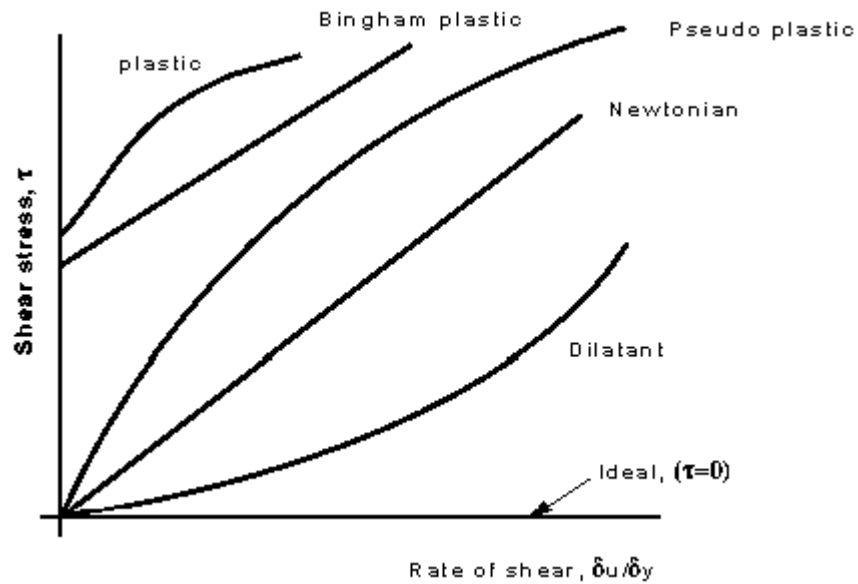
As you will have seen when looking at properties of solids, when the elastic limit is reached they seem to flow. They become plastic. They still do **not** meet the definition of true fluids as they will only flow after a certain minimum shear stress is attained.

Newtonian / Non-Newtonian Fluids

Even among fluids which are accepted as fluids there can be wide differences in behaviour under stress. Fluids obeying Newton's law where the value of μ is constant are known as **Newtonian** fluids. If μ is constant the shear stress is linearly dependent on velocity gradient. This is true for most common fluids.

Fluids in which the value of μ is not constant are known as **non-Newtonian** fluids. There are several categories of these, and they are outlined briefly below.

These categories are based on the relationship between shear stress and the velocity gradient (rate of shear strain) in the fluid. These relationships can be seen in the graph below for several categories



Shear stress vs. Rate of shear strain $\delta u/\delta y$

Each of these lines can be represented by the equation

$$\tau = A + B \left(\frac{\delta u}{\delta y} \right)^n$$

where A, B and n are constants. For Newtonian fluids $A = 0$, $B = \mu$ and $n = 1$.

Below are brief description of the physical properties of the several categories:

- ✚ *Plastic*: Shear stress must reach a certain minimum before flow commences.
- ✚ *Bingham plastic*: As with the plastic above a minimum shear stress must be achieved. With this classification $n = 1$. An example is sewage sludge.
- ✚ *Pseudo-plastic*: No minimum shear stress necessary and the viscosity decreases with rate of shear, e.g. colloidal substances like clay, milk and cement.
- ✚ *Dilatant substances*: Viscosity increases with rate of shear e.g. quicksand.
- ✚ *Thixotropic substances*: Viscosity decreases with length of time shear force is applied e.g. thixotropic jelly paints.
- ✚ *Rheopectic substances*: Viscosity increases with length of time shear force is applied

- ✚ *Viscoelastic materials*: Similar to Newtonian but if there is a sudden large change in shear they behave like plastic.

There is also one more - which is not real, it does not exist - known as the **ideal fluid**. This is a fluid which is assumed to have no viscosity. This is a useful concept when theoretical solutions are being considered - it does help achieve some practically useful solutions.

Liquids and Gasses

Although liquids and gasses behave in much the same way and share many similar characteristics, they also possess distinct characteristics of their own. Specifically

- ✚ A liquid is difficult to compress and often regarded as being incompressible. A gas is easily to compress and usually treated as such - it changes volume with pressure.
- ✚ A given mass of liquid occupies a given volume and will occupy the container it is in and form a free surface (if the container is of a larger volume). A gas has no fixed volume, it changes volume to expand to fill the containing vessel. It will completely fill the vessel so no free surface is formed.

Source : <http://nprcet.org/e%20content/mech/FMM.pdf>