Delayed coking



Photograph of a 4-drum delayed coking unit in a Benica petroleum refinery in California. Source: Valero Energy Corporation



Source: U.S. Securities Exchange Commission A four-drum delayed coking unit. The structures above each pair of two drums contains the drill stems for removing the product coke. The two large, vertical white pipes in the foreground are furnace flue gas stacks.

Delayed coking is one of the <u>chemical engineering</u> unit processes used in many <u>petroleum refineries</u>. The main objective of the delayed coking unit is to convert low value residual products to lighter products of higher value and to produce a coke product.

In brief, the process heats the residual oil from the vacuum distillation unit in a petroleum refinery to its thermal <u>crackingtemperature</u> in the heat transfer tubes of a furnace. This partially vaporizes the residual oil and initiates cracking of the long chain<u>hydrocarbon molecules</u> of the residual oil into hydrocarbon <u>gases</u>, coker naphtha, coker gas oil and petroleum coke. The heater effluent discharges into very large vertical vessels (called "coke drums") where the cracking reactions continue to completion, forming solid petroleum coke which deposits out and accumulates in the coke drums from which the product coke is subsequently removed.^{[11][2][3]} The adjacent photograph depicts a delayed coking unit with four coke drums (two pairs of two drums). However, larger units may have as many as eight drums (four pairs of two drums), each of which may have diameters of up to ten meters and overall heights of up to 43 meters.^[4]

The yield of coke from the delayed coking process ranges from about 18 to 30 percent by weight of the feedstock residual oil, depending the composition of the feedstock and the operating variables. Many refineries world-wide produce as much as 2000 to 3000 tons per day of <u>petroleum</u> coke and some produce even more.^[5] Globally, the total amount petroleum coke produced in 2010 was about 123,000,000 metric tons (123 Mt) and is expected to increase at an annual rate of about 5.6 percent.^[6]

Petroleum coke may also be produced in an oil refinery unit process that utilizes fluidized bed technology.^[7]However, there are very few such facilities in operation and the amount of petroleum coke produced via such technology is virtually insignificant.

Another type of coke, commonly referred to as "metallurgical coke", is the solid carbonaceous material derived from the destructive distillation of low-ash, low-sulfur bituminous <u>coal</u>. <u>Volatile</u> constituents of the coal are driven off by baking in an airless oven at temperatures as high as about 1,200 degrees Celsius (about 2,200 degrees Fahrenheit). Metallurgical coke is used as fuel and as a reducing agent in the <u>iron</u> and<u>steel</u> manufacturing industries. The worldwide consumption of metallurgical coke was about 450,000,000 metric tons (450 Mt) in in 2010.^[8]

Flow diagram and process description

The schematic process flow diagram and description in this section are based on a typical delayed coking unit with two coke drums. However, as mentioned above, larger units may have as many as four pairs of drums (eight drums in total) as well as a furnace for each pair of coke drums.

Typical schematic flow diagram



Process description

Residual oil from the vacuum distillation unit (sometimes including high-boiling oils from other sources within the refinery) is pumped into the bottom of the distillation column called the *main fractionator*. From there it is pumped, along with some injected steam, into the fuel-fired furnace and heated to its thermal cracking temperature of about 480 °C (900 °F). Thermal cracking begins in the pipe between the furnace and the coke drums, and finishes in the coke drum that is on-stream. The injected steam helps to minimize the deposition of coke within the furnace tubes.

Pumping the incoming residual oil into the bottom of the main fractionator, rather than directly into the furnace, preheats the residual oil by having it contact the hot vapors in the bottom of the fractionator. At the same time, some of the hot vapors condense into a high-boiling liquid which recycles back into the furnace along with the hot residual oil.

As cracking takes place in the drum, gas oil and lighter components are generated as a vapor phase and separate from the liquid and solids. The drum effluent is vapor (except for any liquid or solids entrainment) and is directed to main fractionator where it is separated into the desired boiling point fractions.

The solid coke, formed in the on-stream coke drum as the cracking reaction continues to completion, is deposited and remains in the coke drum in a porous structure that allows flow through the pores. Depending upon the overall coke drum cycle being used, a coke drum may fill in 16 to 24 hours.

After the drum is full of the solidified coke, the hot mixture from the furnace is switched to the second drum. While the second drum is filling, the full drum is steamed out to reduce the hydrocarbon content of the petroleum coke, and then quenched with water to cool it. The top and bottom heads of the full coke drum are removed, and the solid petroleum coke is then cut from the coke drum with a high pressure water nozzle, where it falls into a pit, pad, or sluiceway for reclamation to storage.

Typical 24-hour coke drum cycle

There are many coke drum cycles, varying from 12 to 24 hours. However, the one just below is typical of a 24-hour cycle:^{[1][9]}

24-Hour Delayed Coking Cycle		
Step		
On-stream drum fills with coke	24.0	
Switch feed from full drum to empty drum	0.5	
Steam-out of the full drum	2.0	
Cool the full drum by filling with water		
Drain water out of full drum	2.0	
Remove top and bottom heads of full drum		
Drill pilot hole from top to bottom of coke using high-pressure water		
Cut coke out completely also using high-pressure water	4.0	
Replace top and bottom heads of the empty drum		
Steam test and purge to remove air from empty drum		
Warm-up by routing hot vapors from on-line drum into top of empty drum		
Open vapor valve completely and switch feed into empty drum		
Contingency time to allow for slippage in above steps	1.5	
Total overall time	48.0	

Petroleum coke specifications

Typical Specifications for Sponge Coke		
Component	Green coke as produced	Coke calcined at 1300 °C
Fixed carbon, wt %	86 – 92	99.5
Sulfur, wt % ^[9]	2.5 - 5.5	1.7 - 3.0
∨olatile matter, wt %	8- 14	0.5
Moisture, wt %	6-14	0.1
Ash, wt %	0.25	0.4
Iron	0.01	0.02
Nickel	0.02	0.03
Silicon	0.02	0.02
Vanadium	0.02	0.03

The adjacent table lists the specifications for raw petroleum coke produced in a delayed coker and the corresponding specification after the raw coke has been calcined in a rotary kiln at 1200 to 1400 °C (2200 to 2550 °F). The raw coke is classified as *sponge coke* and often referred to as *green coke*^{[10][11][12]}.

Depending on the feedstock composition and certain operating variables, other forms of coke may be produced such as *shot coke* and *needle coke*. However, *sponge coke* is the form most typically formed and this article does not discuss the other forms.



Uses of petroleum coke

The product coke from a delayed coker has many commercial uses and applications.^{[13][14][15]} The largest use is as a fuel. The uses for green coke are:

- As fuel for space-heaters, large industrial steam generators, fluidized bed steam generators, Integrated Gasification Combined Cycle (IGCC) units and <u>cement kilns</u>.
- In silicon carbide foundries.
- For producing blast furnace coke.

The uses for calcined coke are:

- As anodes in the production of aluminum.
- In the production of titanium dioxide.
- As a carbon raiser in <u>cast iron and steel manufacture</u>.
- For producing graphite electrodes and other graphite products such as graphite brushes used in electrical equipment.
- In carbon structural materials.

History

Petroleum coke was first made in the 1860's in the early oil refineries in Pennsylvania which boiled oil in small, iron distillation stills to recover kerosene, a much needed lamp oil. The stills were heated by wood or coal fires built underneath them, which over-heated and coked the oil near the bottom of the stills. After the distillation was completed, the still was allowed to cool and workmen could then dig out the coke and tar.^[15]

In 1913, William Merriam Burton, working as a chemist for the Standard Oil of Indiana refinery at Whiting, Indiana, was granted a patent^[16] for the Burton thermal cracking process that he had developed. He was later to become the president of Standard Oil of Indiana before he retired.

In 1929, based on the Burton thermal cracking process, Standard Oil of Indiana built the first delayed coker. It required very arduous manual decoking.^[15]

In the late 1930's, Shell Oil developed hydraulic decoking using high-pressure water at their refinery in Wood River, Illinois. That made it possible, by having two coke drums, for delayed decoking to become a semi-continuous process.^[15]

From 1955 onwards, the growth in the use of delayed coking increased. As of January 2003, there were 133 petroleum refineries worldwide producing 172,000 tons per day of petroleum coke.^[17] Included in those worldwide data, about 59 coking units were operating in the United States and producing 114,000 short tons per day of coke.^[17]

Other processes for producing petroleum coke

There are other petroleum refining processes for producing petroleum coke, namely the Fluid Coking and Flexicoking processes,^{[18][19]} both of which were developed and are licensed by Exxon Mobil Research and Engineering. The first commercial unit went into operation in 1955. However, neither of those two process achieved widespread, global usage.

There are other similar coking processes, but they do not produce petroleum coke. For example, the Lurgi-VZK Flash Coker which produces coke by the pyrolysis of <u>biomass</u>.^[20]

As noted above, another form of coke (commonly referred to as metallurgical coke) is a solid carbonaceous residue derived from lowash, low-sulfur bituminous coal from which the volatile constituents have been driven off by baking in an oven without oxygen at temperatures as high as 1,200 $^{\circ}$ C.

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