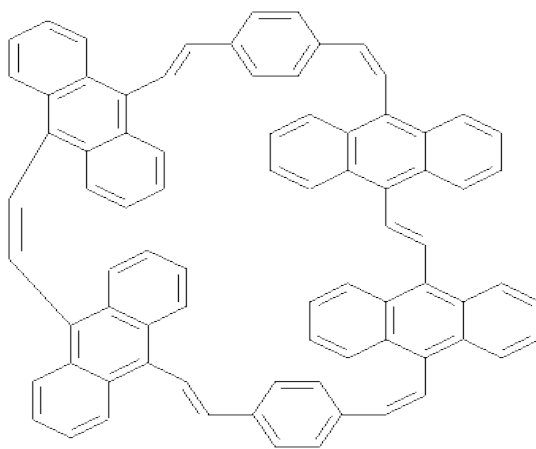


# [36]-ANNULENE

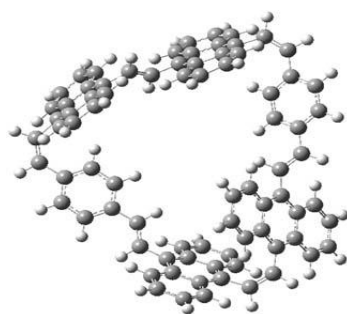
Annulenes can twist, and I have blogged about a number of examples (cations and neutrals). The twist can be portioned into a twist associated with the dihedral angle as one progresses around the compound and writhe which is associated with distortion of the annulene into a third dimension.<sup>1</sup>

Herges has prepared the 36-annulene **1** where the anthracenyl units were introduced to force the loop out of plane.<sup>2</sup> Four different conformations of **1** were isolated by crystallization out of different solvents. **1a** and **1b** were produced in benzene, and they differ by having two half-twists ( $L_k=2$ ) and one half-twist ( $L_k=1$ ), respectively. **1c** was isolated from DMF, with one half-twist. **1d** was isolated from Et<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub>.

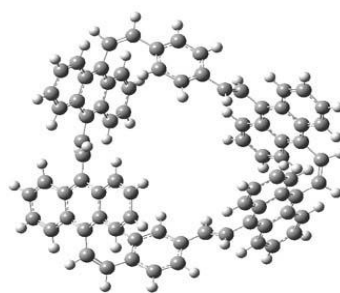


**1**

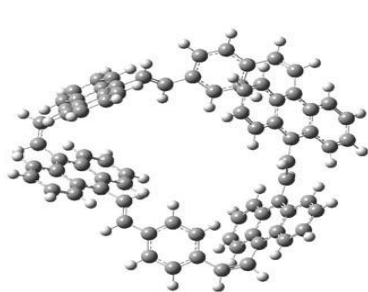
Computations at the B3LYP/6-31g\* level identified 10 low lying conformations, and the ones corresponding to the experimentally observed forms are shown in Figure 1. The computed conformer that matches with **1d** differs from the experimental version by a rotation about one single bond, and the computed version has a different topology than the experiment. One item of note is that all of the 10 computed conformers are dominated by the twist ( $T_w$ ) and have very small writhe.



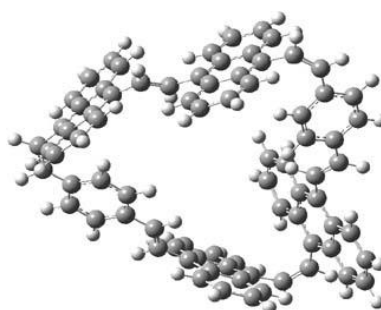
**1a**



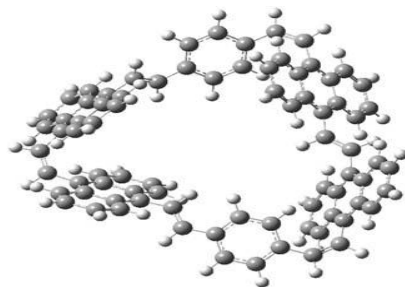
**1b**



**1c**



**1d**



**1e**

**Figure 1.** B3LYP/6-31G\* optimized structures of **1a-e**.

Though not isolated in experiment, one of the low lying conformers has three half-twists ( $L_k=3$ ) and is also shown in Figure 1 as **1e**. Identification of this highly twisted species would be quite interesting.

Source: <http://comporgchem.com/blog/?p=1192>